


## Fertigation and Fertilizer Proportioning

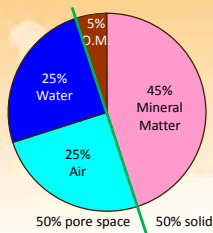
by  
Bob Schultheis  
Natural Resource Engineering Specialist / CPD

*Special thanks to David Trinklein, Division of Plant Sciences,  
University of Missouri, for parts of this presentation*

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***If you take care of your soil, the soil will take care of your plants.***



Component	Percentage
O.M.	5%
Water	25%
Air	25%
Mineral Matter	45%
<b>Total Pore Space (Water + Air)</b>	<b>50%</b>
<b>Total Solid (O.M. + Mineral Matter)</b>	<b>50%</b>

### ***Plant Nutrition vs. Plant Fertilization***

<p><b><u>Nutrition:</u></b> Availability and type of chemical elements in plant</p> 	<p><b><u>Fertilization:</u></b> Adding nutrients to growing medium in proper amounts</p> 
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### ***Why do we still have problems?***

- Focus has been on solving problems
  - Delay crops
  - Reduce quality
  - Lower profits

❖ ***“Need to focus on preventing problems”***



### ***What is Fertigation?***

- Fertilizer + Irrigation = Fertigation
- Nutrient “spoon feeding”
- Can be done by:
  - hand
  - sprinkler system
  - drip irrigation system

### ***Fertigation***

- Advantages
  - Relatively uniform fertilizer applications
  - Flexibility in timing of applications
  - Less fertilizers used
  - Reduced costs
- Disadvantages
  - Potential contamination hazard from equipment malfunctions
  - Backflow prevention devices required
  - Careful handling of liquid fertilizers

### Objectives of Fertigation

- Maximize profit by applying the right amount of water and fertilizer
- Minimize adverse environmental effects by reducing leaching of fertilizers and other chemicals

### Nutrition Affected By

- Chemical considerations
  - pH - water, fertilizer solution
  - Alkalinity - water, fertilizer solution
  - EC - water, fertilizer solution
- Fertilizer analysis
  - Macronutrients, micronutrients
- Non-nutritional elements – possible toxicities
  - Na, Cl, F, Al

### pH

- pH affects the solubility of fertilizers and the efficacy of pesticides and growth regulators
  - The higher the water pH the less soluble these materials are

**pH scale\***  
\*approximate pH numbers

0	1	2	3	4	5	6	7	8	9	10	11	12	13	14
hydrochloric acid	upset stomach	battery acid	sodas	orange juice	tomatoes	bread	pure water	seawater	baking soda	borax	milk of magnesia	ammonia	bleach	sodium hydroxide
	normal stomach acid	vinegar	acidic soil	bananas	potatoes	human saliva	normal rain	eggs	phosphate detergents	antacids	nonphosphate detergents			

### Influence of pH on nutrient availability\*

\*based on a soilless substrate containing sphagnum peat moss, composted pine bark, vermiculite, and sand

Reference: [www.ces.ncsu.edu/depts/hort/hil/hil-558.html](http://www.ces.ncsu.edu/depts/hort/hil/hil-558.html)

### Problems Associated With Improper pH

Low pH	High pH
<ul style="list-style-type: none"> <li>• Toxic:                             <ul style="list-style-type: none"> <li>– Iron</li> <li>– Manganese</li> <li>– Zinc</li> <li>– Copper</li> </ul> </li> <li>• Deficient                             <ul style="list-style-type: none"> <li>– Calcium</li> <li>– Magnesium</li> </ul> </li> <li>• Sensitive                             <ul style="list-style-type: none"> <li>– Ammonium-N</li> </ul> </li> </ul>	<ul style="list-style-type: none"> <li>• Deficient:                             <ul style="list-style-type: none"> <li>– Iron</li> <li>– Manganese</li> <li>– Zinc</li> <li>– Copper</li> <li>– Boron</li> </ul> </li> </ul>

### pH Adjustment

- Raise pH
  - Use fertilizer with lower acid residue
    - ammonium vs. nitrate
    - calcium compounds
  - Apply limestone
    - calcitic --  $\text{CaCO}_3$
    - dolomitic --  $\text{CaMg}(\text{CO}_3)_2$
    - hydrated --  $\text{Ca}(\text{OH})_2$

### *pH Adjustment*

- Lower pH
  - Use fertilizer with acid residue
  - Apply sulfur-containing compounds
 
$$S + O_2 + H_2O \rightarrow H_2SO_4 \rightarrow 2 H^+ + SO_4^{-2}$$
 (requires action of microbes)
  - Sulfuric acid

### GUARANTEED ANALYSIS

NET WEIGHT 25 POUNDS (11.34 KG)

#### PETERS® GENERAL PURPOSE SPECIAL 20-10-20

##### GUARANTEED ANALYSIS

TOTAL NITROGEN (N) ..... 20

12.00% NITRATE NITROGEN

8.00% AMMONIACAL NITROGEN

AVAILABLE PHOSPHORIC ACID (P.O.) ..... 10

SOLUBLE POTASH (K<sub>2</sub>O) ..... 20

Primary Plant Nutrient Sources: Ammonium Nitrate, Ammonium Phosphate, Potassium Nitrate.

Potential Acidity 422 lbs. Calcium Carbonate Equivalent Per Ton.

Manufactured by: Peters® Fertilizer Products, W. R. GRACE &amp; CO., Fogelsville, Pa. 18051

### GUARANTEED ANALYSIS

NET WEIGHT 25 POUNDS (11.34 KG)

#### PETERS® ACID SPECIAL 21-7-7

##### GUARANTEED ANALYSIS

TOTAL NITROGEN (N) ..... 21%

9.05% AMMONIACAL NITROGEN

11.95% UREA NITROGEN

AVAILABLE PHOSPHORIC ACID (P.O.) ..... 7%

SOLUBLE POTASH (K<sub>2</sub>O) ..... 7%

Primary Plant Nutrient Sources: Urea, Ammonium Phosphate, Ammonium Sulfate, Muriate of Potash.

Potential Acidity 1560 lbs. Calcium Carbonate Equivalent Per Ton.

Manufactured by: Peters® Fertilizer Products, W. R. GRACE &amp; CO., Fogelsville, Pa. 18051

### *Conclusions*

- pH greatly affects plant nutrition
- Soilless media prone to pH changes
- Many factors influence pH change
- Monitoring pH important
  - Adjust according to crop and need

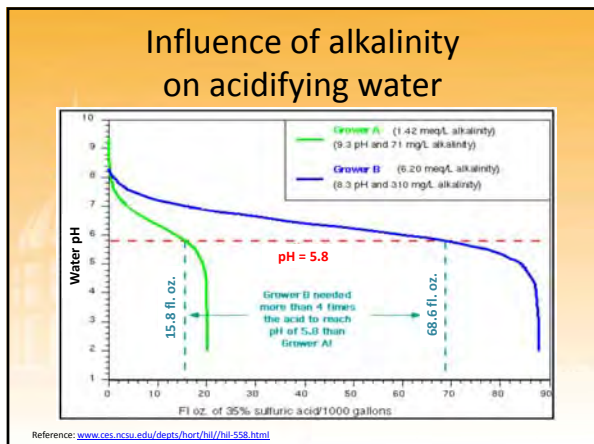
### *Nutrition Affected By*

- Chemical considerations
  - pH - water, fertilizer solution
  - **Alkalinity - water, fertilizer solution**
  - EC - water, fertilizer solution
- Fertilizer analysis
  - Macronutrients, micronutrients
- Non-nutritional elements – possible toxicities
  - Na, Cl, F, Al

### *Alkalinity*

- Alkalinity establishes the buffering capacity of water and affects how much acid is required to change the pH
  - Don't confuse with alkaline pH

Reference: [www.ces.ncsu.edu/depts/hort/hil/hil-558.html](http://www.ces.ncsu.edu/depts/hort/hil/hil-558.html)



### Water Source Quality

**Good**

↑

↓



**Poor**

- Well = check pH & hardness
- Municipal = may be expensive
- Spring = may not be dependable
- River or stream = depends on runoff
- Lake or pond water = sand filters
- Pump to tank on hill = limited use




### Water Quality Analysis

- Inorganic solids = sand, silt
- Organic solids = algae, bacteria, slime
- Dissolved solids (<500 ppm)
  - Iron & Manganese
  - Sulfates & Chlorides
  - Carbonates (calcium)
- pH (5.2-6.8 preferred in greenhouses)
- Hardness (<150 ppm)
- E. coli bacteria

Resources:  
<http://linklablab.missouri.edu/collwater.aspx>  
<https://electropipe.hawaii.edu/Assets/Docs/SP740B.pdf>

### Plugging Potential of Drip Irrigation Systems

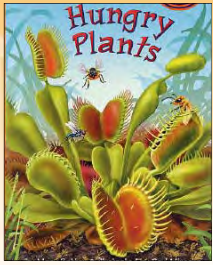
Factor	Moderate (ppm)*	Severe (ppm)*
<b>Physical</b>		
Suspended solids	50-100	>100
<b>Chemical</b>		
pH**	7.0-7.5	>7.5
Dissolved solids	500-2000	>2000
Manganese	0.1-1.5	>1.5
Iron	0.1-1.5	>1.5
Hardness***	150-300	>300
Hydrogen sulfide	0.5-2.0	>2.0

\* ppm = mg/L    \*\* pH is unitless    \*\*\* Hardness: ppm = gpg x 17

### Nutrition Affected By

- Chemical considerations
  - pH - water, fertilizer solution
  - Alkalinity - water, fertilizer solution
  - EC - water, fertilizer solution
- **Fertilizer analysis**
  - Macronutrients, micronutrients
- Non-nutritional elements – possible toxicities
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### How do we actually get the fertilizer to our plants?



### **Application Options**


- Pre-plant
  - Substrate incorporation
- Post-plant
  - Top dress/incorporate
  - Liquid feed

(Might use all three on one crop)




### **Substrate Incorporation**

- Separately
  - Ground limestone (Ca, for pH)
  - Superphosphate (P)
  - Trace elements
  - Slow release materials
- Package
  - “Starter charge” - liquid or granular



### **Fertilizer Types**

- Granular
  - Super phosphate, gypsum
- Slow (controlled) release
  - Osmocote®, MagAmp®
- Water soluble
  - Excel®, Jack’s Classic®
- Organic
  - Bloodmeal, alfalfa meal
- Chelated
  - Sequestrene 330®





### **Slow Release Fertilizers**

- + Extended release period
- + Fewer nutrients leached
- + Use instead of or with liquid feed
- + Form of automation
- Release rate varies
- Affects salts measurement
- Hard to leach excess salts




### **Slow Release--Types**

- Plastic encapsulated
  - Osmocote® (analysis varies)
  - 12-week to 9-month release
- Slowly soluble fertilizers
  - Mag-Amp®
- Sulfur-coated urea
  - Primarily for turf

### **Post-plant (Liquid)**

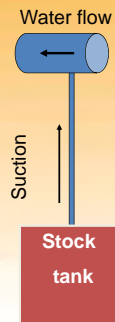
- Most commonly used
- Constant feed (CLF)
  - dilute concentration
  - every watering
- Periodic feed
  - more concentrated
  - intervals (e.g. weekly)






### Venturi Proportioners

- Use pressure differences to draw stock solution into water line
- Pressure changes cause different uptake rate
- Must calibrate for local conditions
  - Water pressure
  - Hose length
- Can require large stock tank



### Estimating Stock Tank Size

- Gallon volume of square or rectangular tank = Length x Width x Depth in feet x 7.5
  - Example: 6' L x 4" W x 2.5' D x 7.5 = 450 gallons
- Gallon volume of round tank (approximate) = Diameter x Diameter x Depth in feet x 6
  - Example: 2' D x 2' D x 3' D x 6 = 72 gallons

### Venturi Proportioner Examples

- Hozon®
  - 1:16 ratio, 35 PSI minimum
  - Unit not more than 50' from hose end
  - Backflow preventer included
  - Do not use with drip irrigation system
  - <http://hozon.com>
- Grow More®
  - 1:16 ratio, 30-90 PSI range
  - Unit not more than 75' from hose end
  - Backflow preventer included
  - Do not use with drip irrigation system
  - <http://www.groworganic.com/siphon-mixer-injector.html>



### Venturi Proportioner Examples

- EZ-Flo®
  - 1:1000 to 1:100 variable ratio (2/3 tsp/gal to 2 TBS/gal)
  - 2 GPM min. flow rate
  - Backflow preventer not included
  - <http://ezfloinjection.com>
- Add-It®
  - 1:200 ratio, 10-80 PSI range
  - 0.5-20 GPM min. flow rate
  - Backflow preventer not included
  - <http://fertilizerdispensers.com/services/add-it.htm>



### Venturi Proportioner Examples

- Young®
  - 1:30 to 1:200 variable ratio
  - 2 GPM min. flow rate
  - Backflow preventer not included
  - Very accurate
  - [http://www.youngproductsinc.com/other\\_products.html](http://www.youngproductsinc.com/other_products.html)

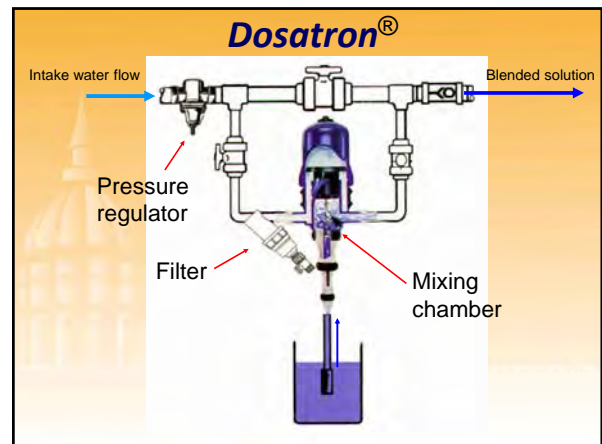
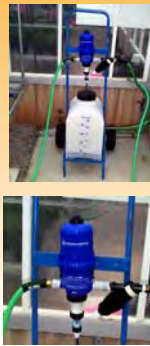


### Positive Displacement

- Flowing water drives piston that pumps stock solution
  - No electricity used
- Rated with min. & max. flow rates depending on model
- Not affected by pressure changes (within range)

### Positive Displacement Examples

- Dosatron® (variable)
  - 1:3000 to 1:4 ratios, 4.3-85 PSI
  - 0.04-14 GPM flow rate
  - Dosing proportional to water flow
  - Operates without electricity, using water pressure as the power source
  - <http://www.dosatron.com>

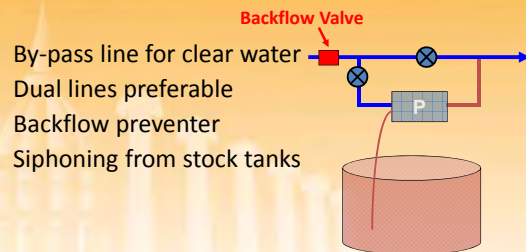


### Positive Displacement Examples

- DosMatic®
  - 1:4000 to 1:10 ratios, 3-100 PSI
  - 0.4-45 GPM flow rate
  - Operates without electricity, using water pressure as the power source
  - <http://hydrosystemsco.com/brands/volumetric-brands/dosmatic/>
- Anderson®
  - <http://www.heanderson.com/products.php>
- Smith®
  - <http://www.usgr.com/fertilization-feeding-injectors/smith-measuremix-injector.php>

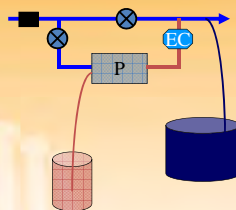


### Proportioner Installation



### Proportioner Calibration

- Check frequently
- < 1:100 : volume uptake vs volume output
- Measure EC of output solution
- In-line EC probe constantly monitors output



### Checking Injector/Calculations

- Check accuracy with salts meter every time new batch of stock is mixed
- Fertilizer companies supply tables of EC values for each of their fertilizers at various concentrations

**20-10-20 peat-lite special**

- 200 ppm = EC of 1.30
- 250 ppm = EC of 1.63
- 300 ppm = EC of 1.95



Reference: <http://www.4oakton.com/iroddetail.asp?parent=28&rodi=3528&e=28&totrec=13>



## Solubility of Selected Fertilizers

### Solubility of Fertilizer in Pure Water, lbs./gal.

Ammonium nitrate	9.8
Calcium nitrate	8.5
Potassium chloride	2.3
Potassium nitrate	1.1

- If two or more fertilizers are to be mixed in the same solution, test their combined solubility by first mixing them in 1-5 gallons of water

Reference: <http://extension.uga.edu/publications/detail.cfm?number=81130>

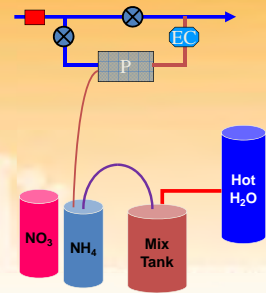
## Stock Mixing

High quality, water soluble materials

Mix in separate tank - pump from another

Best to use warm water when mixing stock - increases solubility

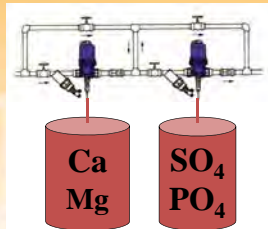
Use separate tanks for different fertilizers



## Stock Mixing Cautions

++ Ca & Mg vs SO<sub>4</sub> & PO<sub>4</sub> --

- High concentrations (>100:1) can cause precipitates
- Precipitates form sludge in tank bottom
- Use two injectors
- Use dual head injector



## Calculations

To determine amount of fertilizer to add to make stock solution:

$$\frac{\text{injector ratio (:1)}}{\% \text{ element}} \times \frac{\text{desired ppm}}{100} \times 1.35$$

= ounces fertilizer/gallon stock

## Calculations

How much fertilizer does one add to a 5 gallon bucket of stock to get 200 ppm N from a 20-10-20 fertilize using a Hozon® injector (1:16)?

$$\frac{16}{20} \times \frac{200}{100} \times 1.35 =$$

$$0.8 \times 2.0 \times 1.35 = 2.16 \text{ oz/gal}$$

$$2.16 \text{ oz/gal} \times 5 \text{ gal} = 10.8 \text{ oz in bucket}$$

## Calculations

How much fertilizer does one add to a 20 gallon tank of stock to get 250 ppm N from a 21-5-19 fertilizer using a Smith® injector (1:100)?

$$\frac{100}{21} \times \frac{250}{100} \times 1.35 =$$

$$4.76 \times 2.5 \times 1.35 = 16.1 \text{ oz/gal}$$

$$16.1 \text{ oz/gal} \times 20 \text{ gal} = 322 \text{ oz}$$

$$322 \text{ oz} / 16 \text{ oz per lb} = 20.1 \text{ lbs fertilizer in tank}$$

### Calculations

How much fertilizer do you add to a 50 gallon tank to get 200 ppm-N from a 15-0-15 fertilizer using a 1:100 injector?

**2 bags +  
45.5 gallons  
water**

**Bags? (25 lbs each)**

$$55.5 / 25 = 2+ \text{ bags}$$

**Set up proportion:**

$$\frac{55 \text{ lbs}}{50 \text{ gal}} = \frac{50 \text{ lbs}}{X \text{ gal}}$$

$$55X = 2500$$

$$X = 45.45 \text{ gallons}$$

### Daily Operations

Which is easier, more efficient and more precise?

55.5 lbs in 50 gallons

2 - 25 lb bags

Weigh out 5.5 lbs from 3rd bag

Fill tank to 50 gal.

50 lbs in 45.5 gallons?

2 - 25 lb bags

Fill tank to 45.5 gal.

**Less mess! No open bags!**

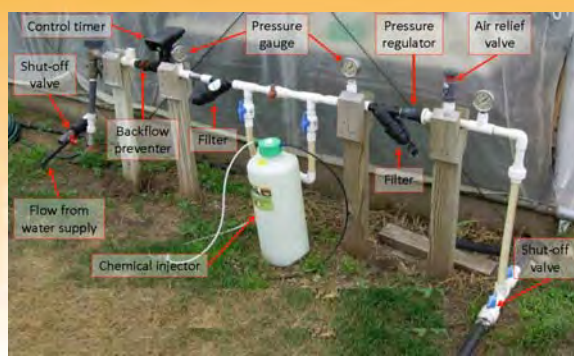
### Fertigation Tips

- Get water supply tested (pH, alkalinity, TDS, etc.)
- Use backflow preventer if required
- Install the injector out of direct sunlight
  - Make sure stock tank is opaque and covered
- Install injector after the timer so tank does not stay under constant pressure
- Always drain unit if there is a chance of freezing
- Be sure fertilizer is 100% water-soluble
  - Make liquid concentrate first from water-soluble powders
  - Strain concentrate to remove undissolved granules

### Fertigation Tips

- Regularly check suction tube filter in stock tank for clogs and holes
- Minimum injection duration of 45-60 minutes is recommended
- Maximum injection duration depends on soil type and nutrient and water requirements of the crop
  - A “reasonable” maximum should not exceed 2 hours per zone

### Drip Irrigation Control Assembly

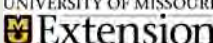


### Conclusion

- Taking a plant from “seed to sale” involves proper fertilization
- There are many ways to get the job done
- The best way is the one that works consistently for you



## Questions??

<p><b>Robert A. (Bob) Schultheis</b> Natural Resource Engineering Specialist Webster County Extension Center 800 S. Marshall St. Marshfield, MO 65706 Voice: 417-859-2044 Fax: 417-468-2086 E-mail: <a href="mailto:schultheisr@missouri.edu">schultheisr@missouri.edu</a> Web: <a href="http://extension.missouri.edu/webster">extension.missouri.edu/webster</a></p> <p><b>UNIVERSITY OF MISSOURI</b>  Extension</p>	<p><b>Program Complaint Information</b> To file a program complaint you may contact any of the following:</p> <p>University of Missouri</p> <ul style="list-style-type: none"><li>▪ MU Extension AA/EEO Office 109 F. Whitten Hall, Columbia, MO 65211</li><li>▪ MU Human Resources Office 130 Heinkel Bldg, Columbia, MO 65211</li></ul> <p>USDA</p> <ul style="list-style-type: none"><li>▪ Office of Civil Rights, Director Room 326-W, Whitten Building 14th and Independence Ave., SW Washington, DC 20250-9410</li></ul>
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